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# 1-(Methoxymethyl)pyrene

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#### **Key indicators**

Single-crystal X-ray study T = 93 KMean  $\sigma(C-C) = 0.001 \text{ Å}$ R factor = 0.046wR factor = 0.142 Data-to-parameter ratio = 21.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The title compound,  $C_{18}H_{14}O$ , crystallizes with aromatic  $\pi$ – $\pi$ stacking interactions.

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#### Comment

With their particular electronic, optical and geometric properties, pyrene and its derivatives (Garcia-Cruz et al., 2004; Takahashi et al., 2004; Borisevich et al., 1992) have attracted considerable recent interest. This is demonstrated in a great number of applications including host-guest (Vögtle, 1996) and metal coordination chemistry (Arrais et al., 2004), as well as fluorescent sensor development (Bren, 2001). Although pyrenes with a functional side arm are important building blocks in this respect, reports on corresponding crystal structures are limited (Foss & Stevens, 1985; Olszak et al., 1989). Here, we describe the structure of 1-(methoxymethyl)pyrene, (I), a new example of this type of compound.

The pyrene part of the molecule shows no significant deviations of bond lengths and angles compared with those of the unsubstituted analogue (Camerman & Trotter, 1965; Allmann, 1970; Hazell et al., 1972; Kai et al., 1978), and is almost planar. The largest deviation from the mean plane through the carbon framework of the pyrene unit is 0.042 (1) Å for atom C1. The torsion angle C2-C1-C17-O1 is 15.1 (1)°, indicating that the methoxymethyl group is turned away from the aromatic plane, while the C-C-O-C fragment itself exhibits a nearly ideal anti-periplanar conformation  $[179.7 (1)^{\circ}]$ .

Due to the aromatic nature of the molecule, the crystal packing of (I) is dominated by arene-arene supramolecular contacts. In contrast to the unsubstituted pyrene, which forms face-to-face dimers packed in a herringbone fashion, the crystal structure of (I) is characterized by molecular stacks which are stabilized by offset face-to-face interactions. Within the stacks, the mean distance between consecutive molecules is 3.5 Å. The molecules of neighboring stacks are inclined to each other at an angle of 86.47 (3)°. Hence the compound has an overall packing structure resembling the pattern found in

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## organic papers

the solid state structures of many aromatic compounds (Desiraju, 1989).

## **Experimental**

The title compound, (I), was synthesized from commercially available pyrene-1-carbaldehyde, which was initially reduced with sodium borohydride in boiling methanol, following an analogous procedure described for the reduction of anthracene-9-carbaldehyde (Steward, 1960), to yield the intermediate compound 1-(hydroxymethyl)pyrene. This was transformed into 1-(bromomethyl)pyrene by the usual bromination with phosphorus tribromide in chloroform (Okamoto *et al.*, 1990). Subsequent treatment with boiling methanol and recrystallization of the product from the same solvent yielded 60% of compound (I) as colorless needles (m.p. 322–323 K). <sup>1</sup>H NMR spectroscopic data (400 MHz, CDCl<sub>3</sub>): δ 8.32 (*d*, ArH, 1H), 8.13 (*m*, ArH, 4H), 8.02 (*m*, ArH, 4H), 5.14 (*s*, CH<sub>2</sub>OH, 2H), 3.49 (*s*, CH<sub>3</sub>, 3H).

#### Crystal data

$C_{18}H_{14}O$	Z = 4
$M_r = 246.29$	$D_x = 1.345 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
a = 4.7220 (10)  Å	$\mu = 0.08 \text{ mm}^{-1}$
b = 20.087 (4)  Å	T = 93 (2)  K
c = 12.824 (3)  Å	Prism, colorless
$\beta = 91.13 \ (3)^{\circ}$	$0.41 \times 0.20 \times 0.17 \text{ mm}$
$V = 1216.1 (5) \text{ Å}^3$	

#### Data collection

Bruker SMART CCD area-detector diffractometer 3712 independent reflections 3712 independent reflections 2920 reflections with  $I > 2\sigma(I)$  Absorption correction: multi-scan (SADABS; Sheldrick, 2000)  $T_{\min} = 0.906, \ T_{\max} = 0.986$   $\theta_{\max} = 30.5^{\circ}$ 

## Refinement

 $\begin{array}{lll} \text{Refinement on } F^2 & w = 1/[\sigma^2(F_{\text{o}}^2) + (0.0988P)^2 \\ R[F^2 > 2\sigma(F^2)] = 0.046 & + 0.1716P] \\ wR(F^2) = 0.142 & \text{where } P = (F_{\text{o}}^2 + 2F_{\text{c}}^2)/3 \\ S = 0.97 & (\Delta/\sigma)_{\text{max}} = 0.001 \\ 3712 \text{ reflections} & \Delta\rho_{\text{max}} = 0.48 \text{ e Å}^{-3} \\ 172 \text{ parameters} & \Delta\rho_{\text{min}} = -0.20 \text{ e Å}^{-3} \\ \text{H-atom parameters constrained} \end{array}$ 

H atoms were positioned geometrically (C-H = 0.95-0.99) and refined as riding, with  $U_{\rm iso}({\rm H})$  = 1.2 or 1.5 times  $U_{\rm eq}({\rm C})$ .

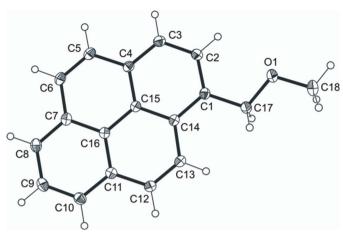
Data collection: *APEX2* (Bruker, 2003); cell refinement: *SAINT* (Bruker, 2003); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2003); software used to prepare material for publication: *SHELXTL*.

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**Figure 1**Perspective view of (I), showing 50% probability displacement ellipsoids for the non-H atoms.

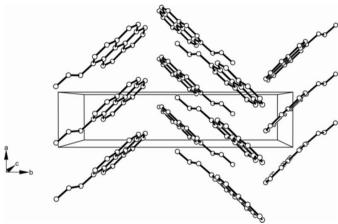


Figure 2 Packing diagram of (I), viewed down the c axis. H atoms have been omitted

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